

Wet Stack Corrosion

A Technical Explanation of the Phenomenon

Wet Stack Corrosion

- ***What is wet stack corrosion?***
- ***What causes wet stack corrosion?***
- ***What are the remedies for wet stack corrosion?***

Wet Stack Corrosion

What is wet stack corrosion?

Under-film corrosion of prepainted products characterized by:

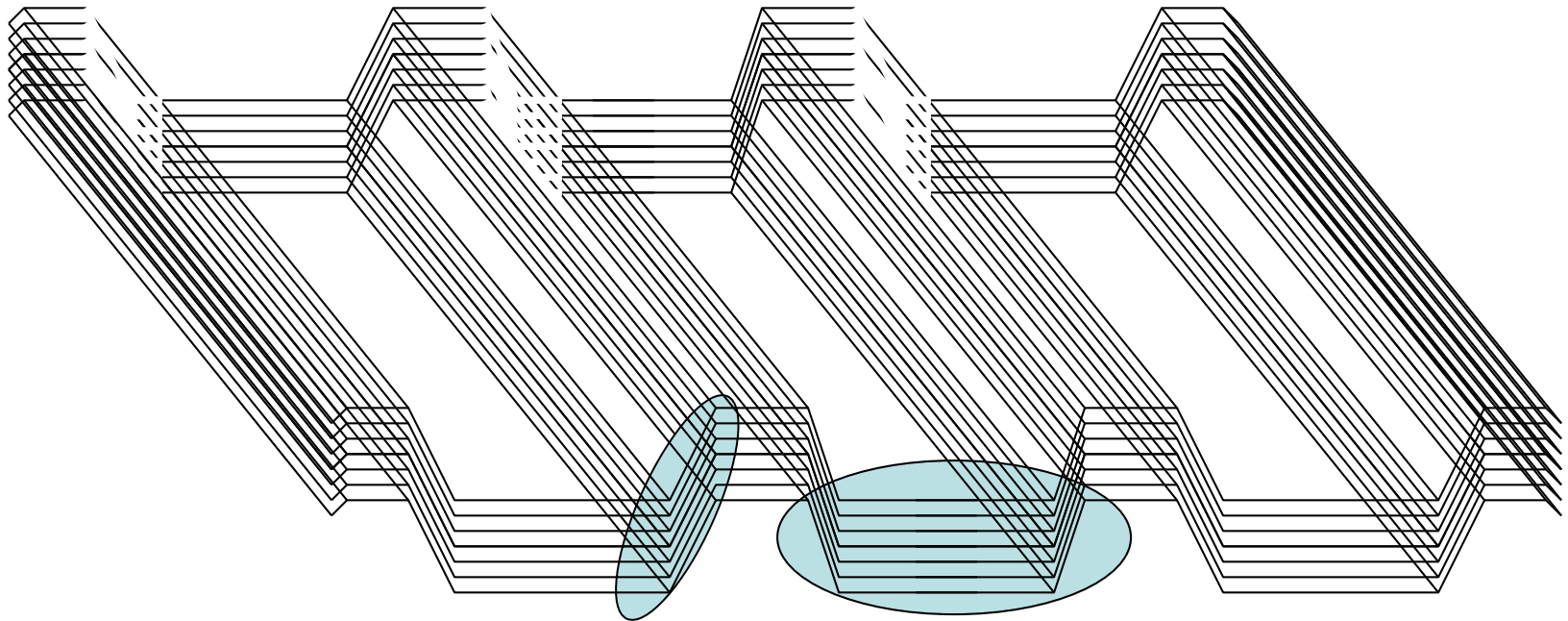
- bottom side and top side panel corrosion initiated beneath the paint film at the substrate surface***
- more severe corrosion on panel ribs and embossments as compared with flat-panel sections***
- no specific pattern of corrosion as related to the panel orientation to the flat stock***

What causes wet stack corrosion?

Consider a typical bundle of panels stored at a job site

General Observations

- Panel ends and sides open to the atmosphere
- Stacking is not perfect leading to offset panel ends
- Panel to Panel contact is greater in rib areas as compared with flat areas
- Panel to Panel gap is smaller in rib areas than in flat areas



Rib slopes and radii have more intimate contact than flat portions of panel

What causes wet stack corrosion?

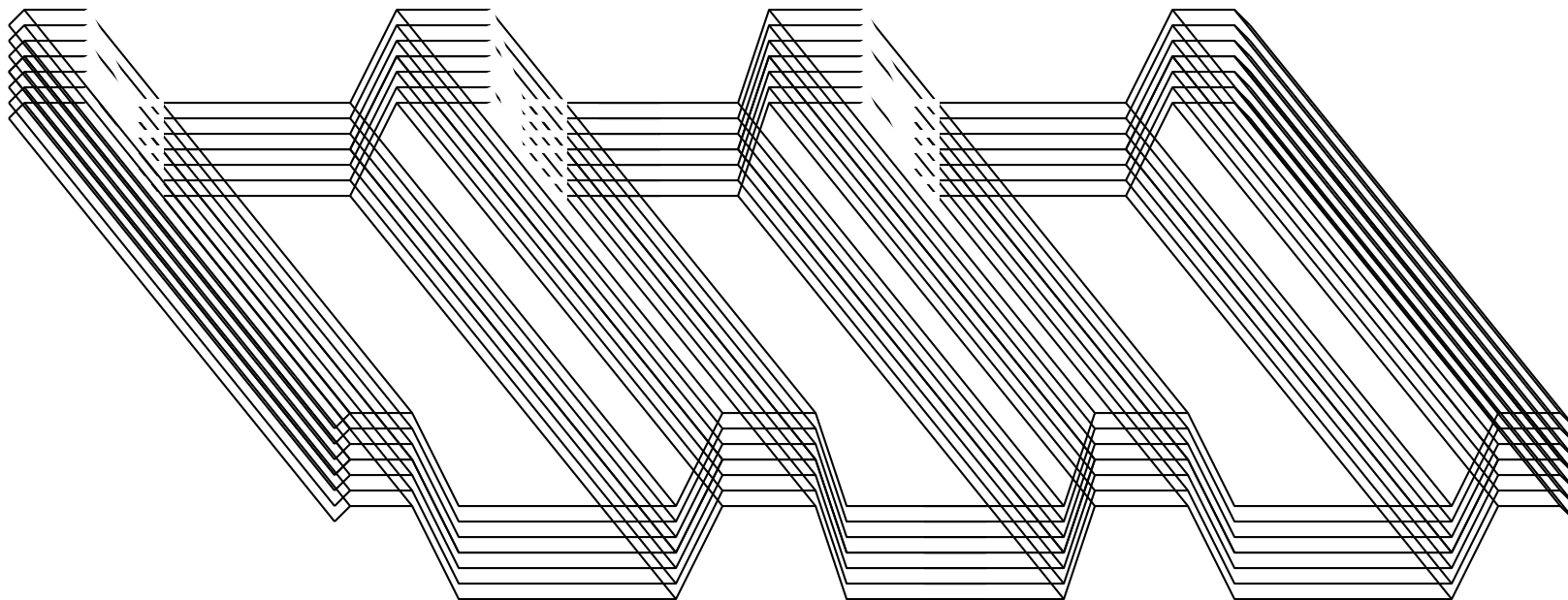
Two sources of water

- Condensation from inter-panel atmosphere
- 'Wicking' of water from wet panel ends

What causes wet stack corrosion?

Condensation of Water from Inter-Panel Atmosphere

- During the normal heating – cooling cycle of a day, if the temperature of the inter-panel atmosphere drops below the dew point, water will condense on to the inter-panel surfaces



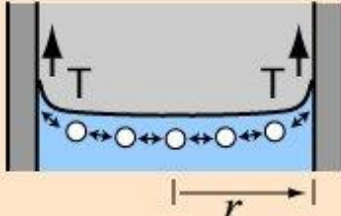
What causes wet stack corrosion?

'Wicking' of Water into Panel Bundles

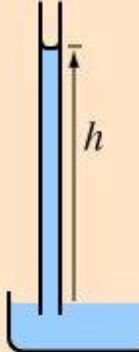
- Occurs when panel edges become wet by dew or rain
- The mechanism for water ingress is 'Capillary Action'

Capillary Action

Capillary action occurs when the adhesion to the walls is stronger than the cohesive forces between the liquid molecules. The height to which capillary action will take water in a uniform circular tube is limited by surface tension. Acting around the circumference, the upward force is

$$F_{upward} = T2\pi r$$


T = surface tension
 ρ = density of liquid



Since it is weight limited it will rise higher in a smaller tube.

The height h to which capillary action will lift water depends upon the weight of water which the surface tension will lift:

$$T2\pi r = \rho g(h\pi r^2)$$

The height to which the liquid can be lifted is given by

$$h = \frac{2T}{\rho r g}$$

[Show calculation](#)

Capillary Movement Calculations

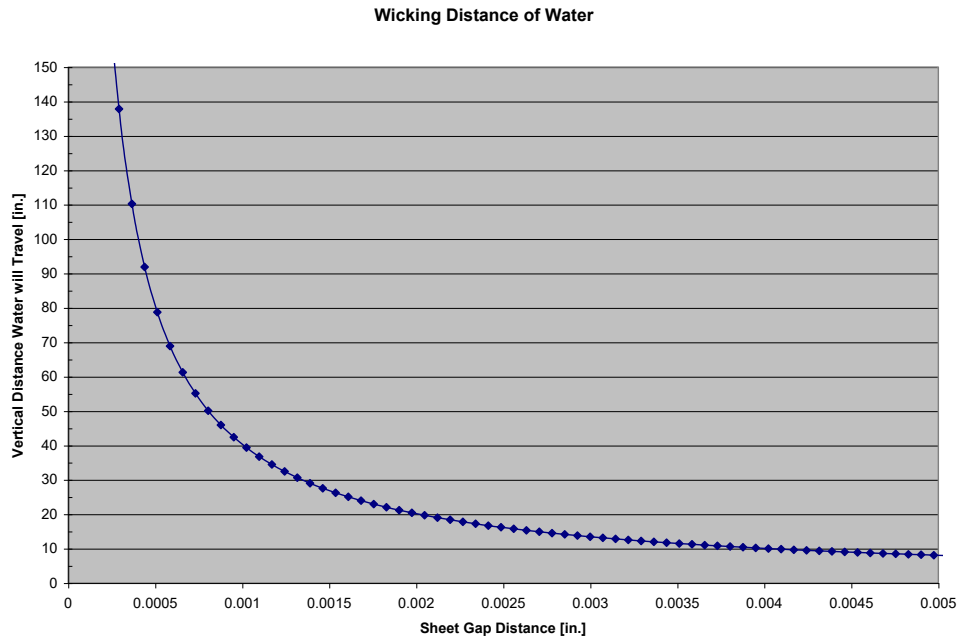
$$h = 2T \cdot \cos(\theta) / (\rho g r)$$

Where: T = surface tension, θ = contact angle of the liquid
 ρ = liquid density, g = acceleration of gravity
 r = $\frac{1}{2}$ the gap distance

Substituting terms for water at sea level we see:

$$h = (1.4 \times 10^{-5}) / r$$

As r decreases water will migrate longer distances.



Capillary Movement of Water

- The distance that water will ingress is directly proportional to (1/inter-panel distance)
- The tighter the stacking, at places where panel contact is not flush, the greater the length of water travel
- Not only will water which ingresses from outside the bundle migrate to areas of tight panel stacking, but also water present in the bundle inter-laps from condensation will also migrate to the areas where the inter-lap distance is less because of the identical capillary effect.

Why does corrosion occur if the material is properly painted?

- **No paint system is totally impervious to water penetration**
- **Paint systems are water resistant but, on a microscopic scale, water moves into and out of the paint film**
- **In normal atmospheres where water can drain and evaporate the amount of water ingress into the paint system is very small**
- **In a panel stack, there is less opportunity for water evaporation and wetness time increases significantly as compared with panels normally exposed to the atmosphere**
- **In a panel stack, the panel edges experience more atmospheric interaction allowing latent water to evaporate to a greater degree thereby decreasing the tendency for wet stack corrosion on panel edges.**

Prevention of Wet Stack Corrosion

- **Proper job site storage**
 - **Bundles should be covered with a water proof tarpaulin or canvas and the bundle bottom should be open to allow for air movement**
 - **Bundles should be stored off of the ground and sloped to allow the runoff of any water that penetrates the tarpaulin**
 - **Bundles should be inspected frequently to ensure that moisture has not penetrated the bundles.**
 - **Bundles should also be protected from any chemical exposure.**
 - **If possible the best storage would be covered storage.**